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Characterisation of Denitrification in the Subsurface Environment of the Manawatū Catchment, New Zealand

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ABSTRACT

A sound understanding of the quantity of nitrate lost from agricultural soils, as well as their transport and transformation in soil-water systems is essential for targeted and effective management and/or mitigation of their impacts on the quality of receiving waters. However, there is currently little known about the occurrence, variability, or factors affecting, nitrate attenuation by subsurface (below the root zone) denitrification in New Zealand, particularly in the Manawatū River catchment. This thesis developed and applied a combination of regional- and local-scale hydrogeochemical surveys and experiments, to gain an insight into the occurrence, variability, and hydrogeological features of subsurface denitrification in the Manawatū River catchment, particularly in the Tararua Groundwater Management Zone (GWMZ).

A regional survey and analysis of samples from 56 groundwater wells conducted in the Tararua GWMZ revealed mainly oxic groundwater with low denitrification potential in the southern part of the catchment (Mangatainoka sub-catchment), whereas mainly anoxic/reduced groundwaters with high potential to denitrify in the middle and northern parts (Upper Manawatū sub-catchments). Oxic groundwaters with enriched nitrate concentrations were generally correlated with coarse textured soil types and aquifer materials (e.g., well-drained soil, gravel rock type), allowing faster movement of percolating water and oxygen diffusion from surface to subsurface environments.

Local-scale laboratory incubations and in-field, push-pull test techniques were evaluated and optimised to measure and quantify denitrification in unsaturated (vadose) and saturated (shallow groundwater) parts of the subsurface environment. A novel incubation technique

using vacuum pouches was found to be more reliable than traditional Erlenmeyer flasks in determining denitrifying enzyme activity (DEA) in subsurface soils (>0.3 m depth) with low denitrification activity. A combination of 75 $\mu\text{g N g}^{-1}$ dry soil and 400 $\mu\text{g C g}^{-1}$ dry soil was also found to provide the optimum DEA in subsurface soils. In the evaluation of the push-pull test, denitrification rates estimated using the measurements of denitrification reactant (nitrate) were found to be significantly higher (6 to 60 times) as compared to the rates estimated using the measurements of denitrification product (nitrous oxide). The estimates of denitrification rates also differed depending on whether a zero-order or first-order kinetic model was assumed. However, either a zero-order or a first-order model appears to be valid to estimate the denitrification rate from push-pull test data.

The optimised laboratory incubation technique and in-field, push-pull test were applied at four sites with contrasting redox properties; Palmerston North, Pahiatua, Woodville, and Dannevirke. The incubation technique revealed that denitrification potential in terms of DEA is highest in the surface soil and generally decreased with soil depth. The push-pull test measured large denitrification rates of 0.04 to 1.07 $\text{mg N L}^{-1} \text{h}^{-1}$ in the reduced groundwaters at depths of 4.5-7.5 m below ground level at two of the sites (Woodville and Palmerston North), whereas there were no clear indications of denitrification in the oxidised shallow groundwaters at the other two sites (Pahiatua and Dannevirke).

This new knowledge, information and techniques advance our scientific capability to assess and map subsurface denitrification potential for targeted and effective land use planning and water quality measures in the Manawatū catchment and other catchments across New Zealand's agricultural landscapes and worldwide.

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TABLE OF CONTENTS

ABSTRACT.....	i
ACKNOWLEDGEMENTS.....	iii
PUBLICATIONS AND PRESENTATIONS.....	v
TABLE OF CONTENTS.....	ix
LIST OF TABLES.....	xiii
LIST OF FIGURES.....	xv
LIST OF ABBREVIATIONS AND SYMBOLS.....	xx
CHAPTER 1.....	1
INTRODUCTION.....	1
1.1 Background.....	1
1.2 Rationale of the Study.....	4
1.3 Study Area.....	6
1.4 Research Objectives.....	7
1.5 Thesis Structure.....	8
CHAPTER 2.....	11
DENITRIFICATION IN THE SUBSURFACE ENVIRONMENT – A REVIEW OF ITS PROCESSES, INFLUENCING FACTORS, MEASUREMENT APPROACHES, AND DENITRIFICATION STUDIES IN NEW ZEALAND.....	11
2.1 Introduction.....	11
2.2 Nitrate and Its Attenuation in the Subsurface Environment.....	11
2.2.1 <i>Nitrate in the subsurface environment</i>	11
2.2.2 <i>Nitrate attenuation processes in the subsurface environment</i>	14
2.2.3 <i>Factors affecting denitrification</i>	18
2.3 Denitrification in the Vadose Zone.....	30
2.3.1 <i>Significance of denitrification in the vadose zone</i>	30
2.3.2 <i>Variability of denitrification in the vadose zone</i>	32
2.3.3 <i>Implications for research</i>	37
2.4 Denitrification in the Saturated Zone.....	38
2.4.1 <i>Saturated zone or groundwater as a pathway of nitrate attenuation and/or contamination</i>	38
2.4.2 <i>Variability of denitrification in the saturated zone (i.e., groundwater)</i>	40
2.4.3 <i>Implications for research</i>	43
2.5 Measuring Denitrification in the Subsurface Environment.....	44
2.5.1 <i>Quantitative measures of denitrification</i>	44
2.5.2 <i>Approaches for measuring denitrification</i>	47
2.5.3 <i>Methods for measuring denitrification in the vadose zone</i>	49

2.5.4	<i>Methods for measuring denitrification below the water table (saturated zone or groundwater)</i>	51
2.5.5	<i>Implications for research</i>	56
2.6	Subsurface Denitrification Studies in New Zealand Environment.....	58
2.6.1	<i>Importance of denitrification studies in New Zealand</i>	58
2.6.2	<i>Subsurface denitrification studies in New Zealand</i>	59
2.6.3	<i>Knowledge gaps and research needs</i>	60
2.7	Concluding Remarks.....	61
CHAPTER 3		64
DENITRIFICATION POTENTIAL IN THE SUBSURFACE ENVIRONMENT IN THE MANAWATŪ RIVER CATCHMENT, NEW ZEALAND: INDICATIONS FROM OXIDATION-REDUCTION CONDITIONS, HYDROGEOLOGICAL FACTORS, AND IMPLICATIONS FOR NUTRIENT MANAGEMENT.....		64
	Abstract.....	64
3.1	Introduction.....	65
3.2	Methods and Materials.....	68
3.2.1	<i>Study area description</i>	68
3.2.2	<i>Groundwater monitoring and quality trend</i>	70
3.2.3	<i>Groundwater survey</i>	71
3.2.4	<i>Groundwater sample collection and analytical methods</i>	72
3.2.5	<i>Data processing for statistical analysis</i>	73
3.2.6	<i>Classification of redox conditions and determination of denitrification potential of groundwater</i>	74
3.2.7	<i>Assessment of effects of hydrochemical processes and hydrogeological factors on denitrification potential</i>	76
3.3	Results and Discussion	77
3.3.1	<i>Groundwater hydrochemistry and distribution of nitrate concentrations and other redox-related parameters</i>	77
3.3.2	<i>Denitrification characteristics of groundwater</i>	82
3.3.3	<i>Hydrochemical processes affecting groundwater chemistry</i>	84
3.3.4	<i>Anthropogenic and hydrogeological factors affecting hydrochemistry and denitrification potential in groundwater</i>	89
3.3.5	<i>Implications of subsurface denitrification characteristics on nitrogen loadings</i>	96
3.4	Conclusions.....	97
CHAPTER 4		99
A NOVEL ACETYLENE INHIBITION INCUBATION TECHNIQUE FOR THE ACCURATE QUANTIFICATION OF DENITRIFYING ENZYME ACTIVITY IN SUBSURFACE SOILS.....		99
	Abstract.....	99
4.1	Introduction.....	100
4.2	Materials and Methods.....	105
4.2.1	<i>Soil sample collection</i>	105
4.2.2	<i>Denitrifying enzyme activity (DEA) measurements</i>	108

4.2.3	<i>Analytical methods and quantification of DEA</i>	113
4.3	Results and Discussion	116
4.3.1	<i>Comparison of DEA measured with different incubation techniques</i>	116
4.3.2	<i>Effect of different substrate amounts on DEA measurements</i>	119
4.4	Conclusions.....	124
CHAPTER 5	126
QUANTIFICATION OF DENITRIFICATION RATE IN SHALLOW GROUNDWATER USING THE SINGLE-WELL, PUSH-PULL TEST TECHNIQUE.....		
Abstract.....		
5.1	Introduction.....	127
5.2	Methods and Materials.....	131
5.2.1	<i>Study area and experimental site</i>	131
5.2.2	<i>Piezometer installation</i>	133
5.2.3	<i>The single-well push-pull tests and analytical methods</i>	134
5.2.4	<i>Quantification of denitrification rate</i>	140
5.3	Results and Discussion	147
5.3.1	<i>Push-pull test results</i>	147
5.3.2	<i>Zero-order denitrification rates</i>	153
5.3.3	<i>First-order denitrification rates</i>	163
5.3.4	<i>Comparison of denitrification rates obtained using denitrification reactant (NO_3^--N) and product (N_2O-N)</i>	172
5.3.5	<i>Comparison of denitrification rates obtained by zero-order and first-order kinetic models</i>	174
5.4	Conclusions.....	179
CHAPTER 6	181
CONTRASTING DENITRIFICATION CHARACTERISTICS IN THE VADOSE ZONE AND SHALLOW GROUNDWATER BENEATH TEMPERATE PASTURE LANDS AND THEIR IMPLICATIONS FOR NITRATE MANAGEMENT IN AGRICULTURAL LANDSCAPES.....		
Abstract.....		
6.1	Introduction.....	182
6.2	Materials and Methods.....	185
6.2.1	<i>Study area and selection of study sites</i>	185
6.2.2	<i>Installation of piezometers</i>	188
6.2.3	<i>Hydrogeologic characteristics of the vadose and saturated zones of the study sites</i>	189
6.2.4	<i>Quantification of denitrification potential in the vadose zone by measuring denitrifying enzyme activity (DEA)</i>	194
6.2.5	<i>Quantification of denitrification rate in the saturated zone by using the single-well, push-pull test technique</i>	197
6.3	Results and Discussion	204
6.3.1	<i>Spatial and temporal variability of nitrate in the vadose zone</i>	204

6.3.2	<i>Spatial and temporal variability of denitrification potential in the vadose zone</i>	204
6.3.3	<i>Spatial and temporal variability of denitrification characteristics in shallow groundwaters</i>	210
6.3.4	<i>Indications from background concentrations of nitrous oxide in groundwater at the study sites</i>	221
6.3.5	<i>Implications of denitrification characteristics on transformation and transport of nitrate in the subsurface environment</i>	224
6.4	Conclusions	228
CHAPTER 7		232
SYNTHESIS AND RECOMMENDATIONS FOR FUTURE WORK		232
7.1	Introduction	232
7.2	Occurrence of and potential for denitrification in the subsurface environment	233
7.3	Quantifying denitrification in the vadose and saturated zones	234
7.3.1	<i>A novel technique for measuring denitrifying enzyme activity in the subsurface soils</i>	235
7.3.2	<i>Quantifying denitrification rate in shallow groundwater using the single-well, push-pull test technique</i>	236
7.4	Variability in denitrification characteristics in the subsurface environment and the contributing factors	238
7.4.1	<i>Spatial and temporal variability</i>	239
7.4.2	<i>Contributing factors</i>	240
7.5	Implications for land management to mitigate nitrate contamination of freshwater resources	242
7.6	Recommendations for future work	243
REFERENCES		246
APPENDICES		274
Appendix A. Summary of Approaches for Measuring Denitrification		275
Appendix B. Summary of In Situ Groundwater Denitrification Rate Measurement Techniques		278
Appendix C. Summary of denitrification studies in New Zealand		283
Appendix D. Supplementary Results		293
Appendix E. Characteristics of the Study Sites		303
Appendix F. Statement of Contribution to Doctoral Thesis Containing Publications		307

LIST OF TABLES

Table 3.1 Threshold concentrations used for identifying redox process in groundwater (modified from McMahon and Chapelle, 2008).....	75
Table 3.2 Summary statistics of groundwater quality parameters measured for groundwater samples collected in the Tararua GWMZ during February-March, 2014.....	79
Table 3.3 Loadings (rotated) from Principal Components Analysis of groundwater quality data from wells in the Tararua GWMZ during February-March, 2014.....	85
Table 3.4 Redox sensitive parameters in groundwater with respect to different soil texture and rock types in the Tararua GWMZ during February-March, 2014.....	92
Table 3.5 Estimated average N leaching and river loading rates in the Upper Manawatū and Mangatainoka sub-catchments in the Tararua GWMZ.....	96
Table 4.1 Physical and chemical properties of soils collected at Massey Dairy No. 1 farm, Palmerston North, New Zealand.....	107
Table 4.2 Denitrifying enzyme activity (DEA) values ($\mu\text{g N}_2\text{O-N kg}^{-1}$ dry soil day^{-1} , mean \pm stdev) for the Manawatū sandy loam soil (sampled in November 2014) at the Massey No. 1 Dairy Farm, Palmerston North, New Zealand.....	120
Table 4.3 Denitrifying enzyme activity (DEA) values ($\mu\text{g N}_2\text{O-N kg}^{-1}$ dry soil day^{-1} , mean \pm stdev) for the Rangitikei silt loam soil (sampled in June 2013) at the Massey No. 1 Dairy Farm, Palmerston North, New Zealand.....	121
Table 5.1 Parameters for the push-pull tests conducted at the two study sites in the Manawatū catchment, New Zealand.....	136
Table 5.2 Summary of different models for estimating zero-order (r) or first-order (k) denitrification rate from single-well push-pull test data.....	143
Table 5.3 Background characteristics of shallow groundwater measured during the assessment of push-pull tests conducted at the two study sites in the Manawatū catchment, New Zealand.....	149
Table 5.4 Estimates of denitrification rate by different models based on zero-order and first order-kinetics using denitrification reactant (nitrate-nitrogen) measured during the push-pull tests conducted at the two study sites in the Manawatū catchment, New Zealand.....	160

Table 5.5 Estimates of denitrification rate by different models based on zero-order and first order-kinetics using denitrification product (nitrous oxide-nitrogen) measured during the push-pull tests conducted at the Palmerston North (PNR) and Woodville (WDV) sites in the Manawatū catchment, New Zealand.....	161
Table 5.6 Comparison of different models assuming zero-order or first-order kinetics for the push-pull tests conducted at Massey No. 1 Dairy Farm, Palmerston North (PNR site) and at a sheep and beef farm near Woodville (WDV site).	178
Table 6.1 Characteristics of the study sites in the Manawatū River catchment.	188
Table 6.2 Physical characteristics of piezometers used for push-pull tests conducted to quantify the denitrification rate at four sites in the Manawatū River catchment, New Zealand.	192
Table 6.3 Physical and chemical characteristics of the soil profile (vadose zone) of the four study sites in the Manawatū River catchment.	193
Table 6.4 Shallow groundwater conditions, push-pull test parameters and measured denitrification rates in Summer/Autumn season at the four sites in the Manawatū River catchment, New Zealand.	199
Table 6.5 Shallow groundwater conditions, push-pull test parameters, and denitrification rates measured in Winter/Spring at the four sites in the Manawatū River catchment, New Zealand.	200
Table 6.6 Denitrifying enzyme activity (DEA) (mean \pm stdev) measured for different soil depths at the four study sites in the Manawatū River catchment, New Zealand.	206

LIST OF FIGURES

Figure 2.1 The nitrogen cycle and its influence upon the water environment.....	13
Figure 2.2 Factors affecting denitrification in soils.	19
Figure 3.1 (a) The Manawatū River catchment and the Tararua GWMZ and its subcatchments and, (b) distribution of groundwater types based on groundwater survey in the Tararua GWMZ during February-March, 2014.....	69
Figure 3.2 Distribution of (a) nitrate-N (mg L^{-1}), (b) dissolved oxygen (mg L^{-1}), (c) dissolved organic carbon (mg L^{-1}) and (d) ferrous iron (mg L^{-1}), in the Tararua GWMZ during February-March, 2014.	80
Figure 3.3 Distribution of (a) ambient redox processes, and (b) wells with high denitrification potential in groundwater in the Tararua GWMZ during February-March, 2014.	83
Figure 3.4 Estimated spatial distribution of contours with iso-factor scores for PCA Component 2 (redox processes in reducing conditions) for groundwater in the Tararua GWMZ based on 56 samples collected in February-March, 2014.....	88
Figure 3.5 Scatter plot of PCA Components 1 and 2 (redox processes in reducing conditions), with respect to identified redox status for groundwater in the Tararua GWMZ based on 56 samples collected in February-March, 2014. Labels in the plot are well numbers.	89
Figure 3.6 Scatter plot of PCA Components 1 and 2 (redox processes in reducing conditions), with respect to (a) land use, (b) soil texture, (c) soil drainage class, and (d) top rock types for groundwater in the Tararua GWMZ based on 56 samples collected in February-March, 2014. Labels in the plot are well numbers.....	91
Figure 3.7 (a) DO and (b) nitrate-N concentrations in groundwater under different combinations of rock type+drainage class in the Tararua GWMZ during February-March, 2014. The absence of box plot for some combinations indicates that no sample fell into such category. Rock types: Al – alluvium, Gr – gravel, Lo – loess.	95
Figure 3.8 (a) Nitrate-N and (b) dissolved organic carbon (DOC) concentrations in groundwater under different combinations of land use+rock type+drainage class in the Tararua GWMZ during February-March, 2014. The absence of	

box plot for some combinations indicates that no sample fell into such category. Rock types: Al – alluvium, Gr – gravel, Lo – loess.	95
Figure 4.1 Laboratory incubation techniques used to measure denitrification enzyme activity (DEA): (a) Erlenmeyer flask, and (b) Vacuum pouches.....	108
Figure 5.1 The Manawatū River catchment showing the location of the study sites: (1) Massey University’s No. 1 Dairy Farm in Palmerston North (PNR site), (2) sheep and beef farm near Woodville (WDV site).....	132
Figure 5.2 Concentrations of (1) nitrate-N and bromide, and (2) nitrous oxide-N during the push-pull test conducted in (a) October 2013, (b) May 2014, and (c) July 2014 at the Massey No. 1 Dairy Farm, Palmerston North, New Zealand (PNR site). Error bars represent standard deviations of the sampling replicates. Volume of test solution – 40 L (Oct 2013) and 100 L (May 2014 and July 2014).	150
Figure 5.3 Concentrations of (1) nitrate-N and bromide, and (2) nitrous oxide-N during the push-pull test conducted in (a) February 2015 and (b) August 2015 at a sheep and beef farm near Woodville, New Zealand (WDV site). Error bars represent standard deviations of the sampling replicates. Volume of test solution 100 L.	151
Figure 5.4 Concentrations of different forms of nitrogen in groundwater samples collected during the push-pull test conducted in the May 2014 at the Massey No. 1 Dairy Farm, Palmerston North, New Zealand (PNR site).	152
Figure 5.5 Nitrous oxide concentrations (uncorrected for dilution) during the push-pull tests conducted in May 2014 (with acetylene) and July 2014 (without acetylene) at the Massey No. 1 Dairy Farm, Palmerston North, New Zealand. Error bars represent standard deviations of the sampling replicates. The volume of test solution for both tests was 100 L.	153
Figure 5.6 Estimating zero-order denitrification rate based on linear regression of dilution-corrected nitrate-N concentrations for the push-pull tests conducted in October 2013, May 2014, and July 2014 at the Massey No. 1 Dairy Farm, Palmerston North, New Zealand (PNR site).....	155
Figure 5.7 Estimating zero-order denitrification rate based on linear regression of dilution-corrected nitrate-N concentrations for the push-pull tests conducted in February 2015 and August 2015 at a sheep and beef farm near Woodville, New Zealand (WDV site).	155

Figure 5.8 Estimating zero-order denitrification rate from a curve fitted on the dilution-corrected nitrate-N concentrations (Trudell et al., 1986) during the push-pull test conducted in October 2013, May 2014, and July 2014 at the Massey No. 1 Dairy Farm, Palmerston North, New Zealand (PNR site).	156
Figure 5.9 Estimating zero-order denitrification rate from a curve fitted on the dilution-corrected nitrate-N concentrations (Trudell et al., 1986) during the push-pull test conducted in February 2015 and August 2015 at a sheep and beef farm near Woodville, New Zealand (WDV site).....	156
Figure 5.10 Estimating zero-order denitrification rate based on Snodgrass and Kitanidis (1998) for the push-pull tests conducted in October 2013, May 2014, and July 2014 at the Massey No. 1 Dairy Farm, Palmerston North, New Zealand (PNR site).....	157
Figure 5.11 Estimating zero-order denitrification rate based on Snodgrass and Kitanidis (1998) for the push-pull tests conducted in February and August 2015 at a sheep and beef farm near Woodville, New Zealand (WDV site).	157
Figure 5.12 Estimating zero-order denitrification rate from a curve fitted on the dilution-corrected nitrous oxide-N concentrations (Sanchez-Perez et al., 2003) for the push-pull tests conducted in (a) October 2013 and (b) May 2014 at the Massey No. 1 Dairy Farm, Palmerston North, New Zealand (PNR site).	158
Figure 5.13 Estimating zero-order denitrification rate from a curve fitted on the dilution-corrected nitrous oxide-N concentrations (Sanchez-Perez et al., 2003) for the push-pull tests conducted in (a) February 2015 and (b) August 2015 at a sheep and beef farm near Woodville, New Zealand (WDV site).	159
Figure 5.14 Estimating first-order denitrification rate based on Haggerty et al. (1998) using the nitrate-N concentrations measured during the push-pull tests conducted in (a) October 2013, (b) May 2014, and (c) July 2014 at the Massey No. 1 Dairy Farm, Palmerston North, New Zealand (PNR site). The data points shown are average values of triplicates.....	166
Figure 5.15 Estimating first-order denitrification rate based on Haggerty et al. (1998) using the nitrate-N concentrations measured during the push-pull tests conducted in (a) February 2015 and (b) August 2015 at a sheep and beef farm near Woodville, New Zealand (WDV site). The data points shown are average values of triplicates.	167

Figure 5.16 Estimating first-order denitrification rate based on Snodgrass and Kitanidis (1998) for the push-pull tests conducted in (a) October 2013, (b) May 2014, and (c) July 2014 at Massey No. 1 Dairy Farm, Palmerston North, New Zealand (PNR site).....	168
Figure 5.17 Estimating first-order denitrification rate based on Snodgrass and Kitanidis (1998) for the push-pull tests conducted in (a) February 2015, and (b) August 2015 at a sheep and beef farm near Woodville, New Zealand (WDV site).....	169
Figure 5.18 Estimating first-order denitrification rate based on Korom et al. (2005, 2012) for the push-pull tests conducted in (a) October 2013, (b) May 2014, and (c) July 2014 at Massey No. 1 Dairy Farm, Palmerston North, New Zealand (PNR site).....	170
Figure 5.19 Estimating first-order denitrification rate based on Korom et al. (2005, 2012) for the push-pull tests conducted in (a) February 2015 and (b) August 2015 at a sheep and beef farm near Woodville, New Zealand (WDV site).	171
Figure 6.1 Locations of the study sites at 1) Pahiatua, 2) Woodville, 3) Dannevirke, 4) Palmerston North in the Manawatū River catchment, New Zealand.....	186
Figure 6.2 Trend in average NO ₃ -N and Br concentrations (excluding background concentrations) measured during the push-pull tests conducted at the Pahiatua (PAH) site in the Manawatū River catchment in (1) the Summer/Autumn 2015 and (2) the Winter/Spring 2015. The time elapsed (hr) was measured from the end of test solution injection. Error bars show the standard deviation for three replicates.	214
Figure 6.3 Trend in average NO ₃ -N and Br concentrations (excluding background concentrations) measured during the push-pull tests conducted at the Woodville (WDV) site in the Manawatū River catchment in (1) the Summer/Autumn 2015 and (2) the Winter/Spring 2015. The time elapsed (hr) was measured from the end of test solution injection. Error bars show the standard deviation for three replicates.	215
Figure 6.4 Trend in average NO ₃ -N and Br concentrations (excluding background concentrations) measured during the push-pull tests conducted at the Dannevirke (DAN) site in the Manawatū River catchment in (1) the Summer/Autumn 2015 and (2) the Winter/Spring 2015. The time elapsed (hr) was measured from the end of test solution injection. Error bars show the standard deviation for three replicates.	216

Figure 6.5 Trend in average $\text{NO}_3\text{-N}$ and Br concentrations (excluding background concentrations) measured during the push-pull tests conducted at the Palmerston North (PNR) site in the Manawatū River catchment in (1) Summer/Autumn 2014 and (2) the Winter/Spring 2013. The time elapsed (hr) was measured from the end of test solution injection. Error bars show the standard deviation for three replicates.216

Figure 6.6 Trend in average nitrous oxide ($\text{N}_2\text{O-N}$) concentrations measured during the push-pull tests conducted at the four study sites in the Manawatū River catchment in (1) the Summer/Autumn and (2) Winter/Spring seasons. The time elapsed (hr) was measured from the end of test solution injection. Error bars show the standard deviation for three replicates.217

LIST OF ABBREVIATIONS AND SYMBOLS

Ag	silver
AI	acetylene inhibition
Al	alluvium
amsl	above mean sea level
ANOVA	Analysis of Variance
APHA	American Public Health Association
BD	bulk density
bgl	below ground level
Br ⁻	bromide
C	carbon
Ca ²⁺	calcium
CBE	charge balance error
Cl ⁻	chloride
CO ₂	carbon dioxide
C ₂ H ₂	acetylene
DAN	Dannevirke
DEA	denitrifying enzyme activity
DEM	digital elevation model
DIC	dissolved inorganic carbon
DNRA	dissimilatory nitrate reduction to ammonium
DO	dissolved oxygen
DOC	dissolved organic carbon
<i>Eh</i>	redox potential
EU	European Union
FeS ₂	pyrite
Fe ²⁺	ferrous iron
FSL	Fundamental Soil Layer
g	gram
GC	gas chromatograph
Gr	gravel
GWMZ	Groundwater Management Zone
h	hour
ha	hectare
HCO ₃ ⁻	bicarbonate
HWC	hot water-extractable carbon
H ₂ O	water
H ₂ SO ₄	sulphuric acid
ICP-OES	Inductively coupled plasma optical emission spectrometry
K ⁺	potassium
KBr	potassium bromide
kg	kilogram
kJ	kilojoule
km	kilometre
km ²	square kilometre
KNO ₃	potassium nitrate
K ₂ Cr ₂ O ₇	potassium dichromate
L	litre

Lo	loess
LSD	Least Significant Difference
m	metre
M	million
MAV	maximum allowable value
mg	milligram
mg L ⁻¹	milligram per litre
Mg ²⁺	magnesium
mL	millilitre
mm	millimetre
Mn ²⁺	manganese
mol	mole
MRT	Mean Residence Time
mV	millivolts
m ³	cubic metre
N	nitrogen
Na ⁺	sodium
NH ₄ ⁺	ammonium
NO	nitric oxide
NO ₂ ⁻	nitrite
NO ₃ ⁻	nitrate
NPSFM	National Policy Statement for Freshwater Management
N ₂	nitrogen gas
N ₂ O	nitrous oxide
OC	organic carbon
ORP	oxidation-reduction potential
O ₂	oxygen
P	phosphorus
PAH	Pahiatua
PCA	Principal Components Analysis
PE	polyethylene
PNR	Palmerston North
ppb	parts per billion
ppm	parts per million
PVC	polyvinyl chloride
rpm	revolution per minute
R ²	coefficient of determination
SO ₄ ²⁻	sulphate
SPC	specific conductance
UK	United Kingdom
WDV	Woodville
WFPS	water-filled pore space
yr	year
°C	degree Celcius
µg	microgram
µL	microlitre
µm	micrometre

