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# Fate of Hydroxyapatite Nano particles during

# In Vitro Gastrointestinal Digestion

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#### **Abstract**

There is an increasing change in population demographics towards an aging population in the world, which had led to the availability of various commercial nutritionally supplemented products. Hydroxyapatite (HA), with chemical formula Ca<sub>10</sub>(PO<sub>4</sub>)<sub>6</sub>(OH)<sub>2</sub>, is an insoluble calcium salt used for calcium supplementation because of its similarity to the minerals found in human bone and teeth. The insoluble calcium salts are preferred over the soluble ones because of their high heat stability during milk processing under high heat treatment. However, the drawback of insoluble calcium salts is the tendency to sediment during storage resulting in unfavourable gritty texture. Thus, reduction in particle sizes into micron to nano-size improves the dispersion of these insoluble salts. However, the application of nano-sized particles in food products have raised concerns from both the regulatory organizations and consumers on the implications related to both the environmental and health safety aspects. Thus, the objective of the study is to determine the digestion behaviour of nano-sized needle/rod shaped HA (nHA) when added into skim milk during in vitro gastrointestinal digestion. Determination of calcium such as soluble and ionic calcium was conducted to determine the dissolution of nHA. The structural changes and the crystallographic changes of nHA were determined using electron microscopy and x-ray diffraction techniques. The results of in vitro gastric digestion showed presence of undissolved nHA particles even after 240 min of gastric and 120 min of intestinal digestion when examined under TEM, while the XRD analysis detected the presence of crystalline nHA in the first 120 min of gastric digestion. Thus, the possible mechanisms leading to the incomplete dissolution of nHA under acidic conditions of the stomach are discussed subsequently.

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## List of abbreviations

°C Degree(s) Celsius

% Percent

**CCP** Colloidal calcium phosphate

CaP Calcium phosphate

**EDS** Energy dispersive spectroscopy

gram(s) g

GI Gastrointestinal

h hour(s)

Hydroxyapatite HA

HGS Human gastric simulator

L Litre(s)

nHA Nano hydroxyapatite particles

nHA blank nHA blank

 $nHA \; _{milk}$ nHA in milk

Milligram(s) mg Minute(s) min

ml Millilitre(s) Millimolar (mmol.L-1)

mmol Millimole(s)

mol Mole(s)

mM

Milk blank Milk blank

Nanometre(s) nm

NMs Nano materials

NPs Nano particles

Recommended dietary allowance **RDA** 

SBF Simulated biological fluid

SGF Simulated gastric fluid

SIF Simulated intestinal fluid

SM Skim milk

TEM Transmission electron microscopy

WP Whey protein

WPI Whey protein isolate

XRD X-ray diffraction

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**Appendix 1** Calculation for addition of nHA in skim milk

**Appendix 2** Determination of dissolution (%) of nHA in nHA blank.

**Appendix 3** Determination of dissolution (%) of nHA in nHA milk.

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