Copyright is owned by the Author of the thesis. Permission is given for a copy to be downloaded by an individual for the purpose of research and private study only. The thesis may not be reproduced elsewhere without the permission of the Author. THE

IDENTIFICATION

OF

NEW ZEALAND OBSIDIANS

A Thesis

presented in partial fulfilment of the requirements for the degree of Master of Science in Chemistry at Massey University

GAYE COILA ARMITAGE

.

1971

ABSTRACT

Obsidian has proved to be a very valuable aid to archaeological investigations, especially in the area of prehistory. Its geological and archaeological occurrence is worldwide.

The aim of this investigation was to establish a satisfactory method of separating the eight known New Zealand sources of flake quality obsidian. Emission spectrography had been used with partial success but the maximum precision of this analytical method is $\pm 5\%$. A more precise analytical method would enable the sources to be separated more readily.

A satisfactory method of sample preparation was developed which gave solutions in which iron, manganese and zinc could be analysed by atomic absorption spectrophotometry, and sodium and potassium by flame photometry.

Replicates of a representative sample from each source were analysed to establish the variation due to the method. This was found to be less than 2%, except when the zinc concentration was below 60 ppm. Analysis of a number of samples from six of the eight sources (Arid Island and Maraetai excluded), established that for these sources the variation in the concentration of at least one of the five elements, was greater between sources than within a source.

All sources, with the possible exception of Taupo and Arid Island, could be distinguished on the basis of the five named elements. Manganese was found to be the most useful element for characterization.

Only one Arid Island sample was available. In all properties it

was very similar to the Taupo pieces analysed, but it may not have been representative. The Maraetai samples available were not flake quality, therefore only one sample was analysed.

The densities of a number of samples from each source were determined by two methods, the hydrostatic weighing method and temperature variation free flotation method. The flotation liquid for the latter was a mixture of bromoform and 1-pentanol.

Successful separation by densities was limited because of extensive overlap between sources. Mayor Island and Kaeo samples were more dense and could be separated from the other six sources, but not from each other. The density method was useful, however, for pieces too small to be analysed chemically. It also had the advantages of being non-destructive and being more readily adaptable to the field.

The two methods of characterization, chemical analysis and density determination, were applied to obsidians from eight archaeological sites located in the following areas, Coromandel, Waikato, Kaipara, Auckland (3), Bluff and Manawatu. More than two hundred and fifty pieces were analysed and more than 95% were positively identified.

ACKNOWLEDGEMENTS

The author wishes to express her thanks to Dr R.D. Reeves for his practical help and invaluable guidance throughout the course of this work. His assistance is gratefully acknowledged.

A special acknowledgement is due to Mr G.K. Ward who generously collected most of the natural source obsidians. Without this material the following study would not have been possible. Mr P. Bellwood, Mr M.T. Timperley and Mr J.E. Armitage are thanked for supplying additional natural source material.

Mr P. Bellwood, Dr R.C. Green, Mr G.S. Park and Mr B.G. McFadgen are thanked for providing the archaeological obsidians which were used to test the characterization methods developed.

The author would like to take this opportunity to thank all members of the Chemistry-Biochemistry Department, especially those of the Inorganic Research Laboratory, for their help and encouragement during this work.

Thanks are also due to Miss J. Quigan for her expert typing of the thesis.

TABLE OF CONTENTS

					Page
ABSTRACT				•••	ii
ACKNO%LEDGEMENTS					iv
TABLE OF CONTENTS	5				v
LIST OF TABLES			.î.		vii
LIST OF FIGURES		•••		•••	viii
CHAPTER I: INTRO	DUCTION				1
1. Occurrer	nce and Use of	Obsidian			1
2. Characte	erization of O	bsidian			2
3. Obsidiar	1 Dating			•••	9
4. New Zeal	land Obsidians		•••	•••	11
(i)	Natural Sourc	es	•••	•••	11
(ii)	Archaeologica	l Usage			14
CHAPTER II: AIMS OF THE PRESENT INVESTIGATION 16					16
CHAPTER III: DEN	ISITY METHODS			•••	22
1. Free Flo	otation Method		•••		22
(i)	Preparation o	f Flotation M	fixture		24
(ii)	Calibration o	of Pycnometer			28
(iii)	Density of Fl Temperatur		d as a Functi	lon of	32
(iv)	Experimental	Procedure for	Density Meas	surements	34

						Page
2.	Hydrostatic	Weighing	Method			36
	(i) Buoy	ancy Corr	ection			36
	(ii) Expe	rimental	Procedure	for Density	Measurements	37
CHAPTER	IV: INVESTIG DENSITY	ATION OF MEASUREME		DURCE OBSIDI.	ANS BY	38
CHAPTER	V: CHEMICAL	ANALYSIS	METHOD			49
1.	Preliminary	Experimen	ts	•••	•••	49
2.	Experimental	Procedur	e	•••	•••	54
CHAPTER	VI: INVESTIG CHEMICAL	ATION OF ANALYSIS			ANS BY	59
1.	Variation Du	e to the	Analytical	L Method		61
2.	Variation in	Chemical	Composit	lon within a	Source	66
	VII: INVESTI		ARCHAEOLO	DGICAL SITE	OBSIDIANS	77
1.	Skipper's Ri	dge	•••	•••	•••	78
2.	Sunde Site		•••	•••	•••	93
3.	Motutapu Und	efended S	ites	•••	•••	95
4.	Hamlins Hill			•••	•••	97
5.	Lake Mangaka	ware		•••	•••	98
6.	Otakanini				•••	99
7.	Foxton Beach	l			•••	103
8.	Tiwai Point		•••			105
CHAPTER	VIII: SUMMAR	Y AND SUG	GESTIONS I	FOR FURTHER	WORK	110
REFERENC			•••	•••	•••	113

vi)

LIST OF TABLES

				Page
I.1.	Analysis of a Mayor Island Ob	sidian	•••	4
II.1.	Ratios of Relative Intensitie with Corresponding Standar Coefficients of Variation	d Deviations		17
III.1.	Volume of Pycnometer as f(T)		•••	29
III.2.	Density of Flotation Liquid a	s f(T)	•••	33
IV.1.	Natural Source Obsidians Exam ments and by Chemical Anal		ty Measure-	39
IV.2.	Natural Source Obsidians Exam	ined by Densi	ty Methods	38
IV.3.	Densities of Natural Source O	bsidians	•••	42
V.1.	Instrumental Conditions for F	e, Mn, Zn	•••	56
VI.1.	Mean Element Concentrations as of Replicate Analyses of S			62
VI.2.	Coefficients of Variation of Single Obsidian Samples	Replicate Ana	lyses of	62
VI.3.	Chemical Analyses of Natural	Source Obsidi	ans	67
VI.4.	Analysis of Striated and Unst. Same Sample, MT/M2	riated Obsidi	an from the	74
VII.1.	Skipper's Ridge Obsidians	•••	•••	79
VII.2.	Summary of Skipper's Ridge Re	sults	•••	91
VII.3.	Sunde Site Obsidians	•••	•••	93
VII.4.	Motutapu Undefended Site Obsi	dians		96
VII.5.	Hamlins Hill Obsidians	•••	•••	97
VII.6.	Lake Mangakaware Obsidians	•••	•••	98
VII.7.	Otakanini Obsidians		•••	99
VII.8.	Foxton Beach Obsidians	•••	•••	103
VII.9.	Tiwai Point Obsidians	•••	•••	106

vii)

LIST OF FIGURES

Between Pages

Fig. 1	Natural Sources of Flake Qua New Zealand	lity Obsidiar	1 in •••	11 - 12
Fig. 2	Volume of Pycnometer as f(T)		•••	28 - 29
Fig. 3	Density of Flotation Liquid	as f(T)		32 - 33
Fig. 4	Sodium in Obsidians	•••	•••	70 - 71
Fig. 5	Potassium in Obsidians	•••	•••	70 - 71
Fig. 6	Iron in Obsidians	•••	•••	70 - 71
Fig. 7	Manganese in Obsidians	•••	•••	70 - 71
Fig. 8	Zinc in Obsidians	•••	•••	70 - 71

viii)